



URBAN SOIL NUTRIENT MANAGEMENT PROGRAM

Presented by the Master Gardeners of Lycoming County

OBJECTIVES

Educate and engage homeowners to:

- •Understand how to minimize the amounts of nutrients and pesticides from reaching our water ways (Chesapeake Bay)
- Gain better understanding of Proper lawn care
- Understand the importance of Soil testing

What is soil?

- Unconsolidated cover on the surface on the Earth.
- Made up of mineral particles, organic matter, air and water.
- Capable of supporting plant growth



Soil horizon/Soil profile



A: mineral horizon w/ organic matter

E: subsurface horizon, little organic matter

B: subsoil

C: unconsolidated material

R: bedrock

Roles of Soil

Soils play multiple roles in the quality of life throughout the world.



Importance of Proper Lawn Care

- Soil protection ground cover
 - ~ prevents erosion
- Nonpoint Source Pollution Reduction
 - ~ correct fertilizing and pesticide use decreases pollution downriver
- Chesapeake Bay Tributary Strategy
 - river-specific 'on-the-ground' actions



PROBLEMS CAUSED BY IMPROPER LAWN MOWING

- Restricts root growth
- Increases susceptibility to damage from insects, disease, drought, traffic, and weed infestation

Optimum mowing heights for lawns:

- Kentucky bluegrass 2.0 to 3.5"
- Perennial ryegrass 2.0 to 3.5"
- Fine fescue 2.0 to 3.5"
- Tall fescue 2.5 to 4.0"



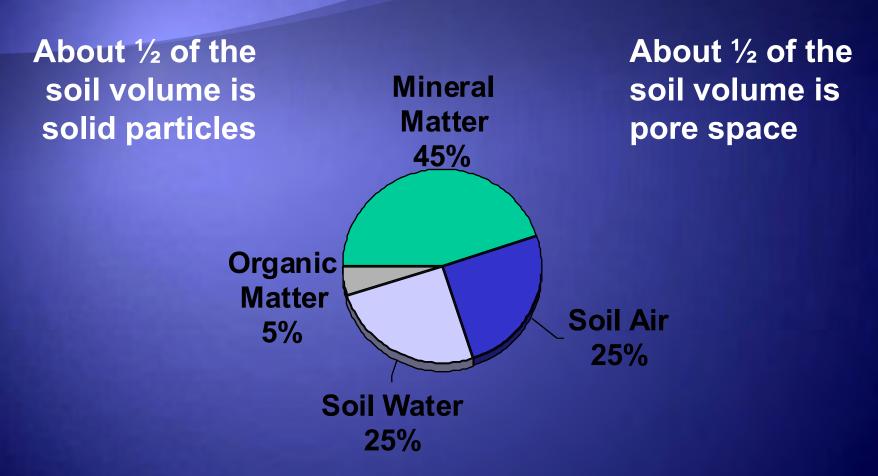
Mineral vs. Organic Soils

- Mineral Soils < 20% organic matter (OM)
- Organic Soils > 20% OM
- Most PA soils contain 1% to 5% OM

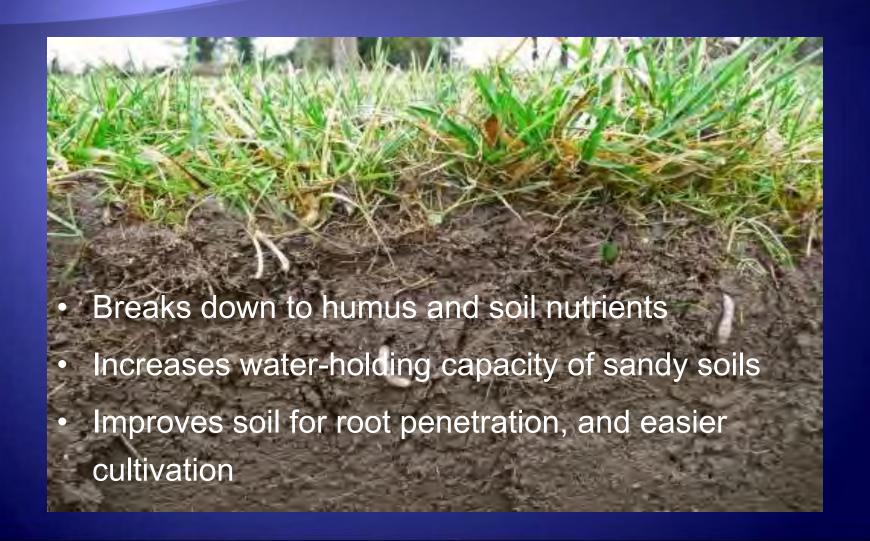


Soil Components

The 4 parts of soil



Organic Matter



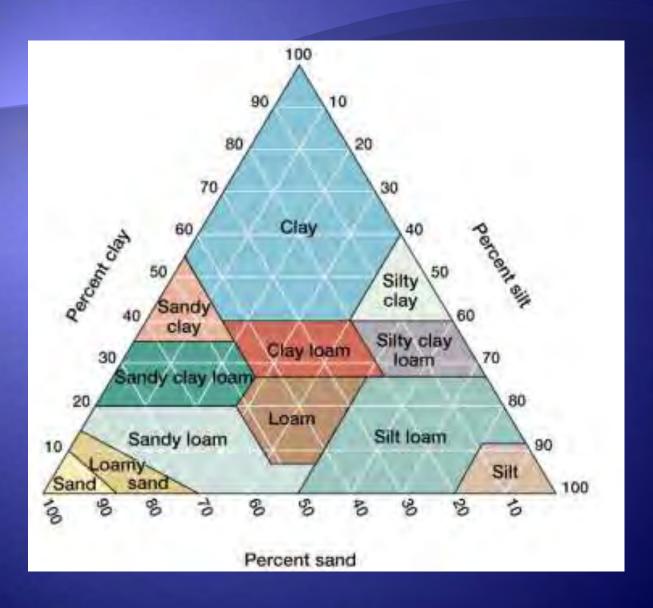
Physical Properties of Soil

- Color
- Texture
- Structure
- Internal drainage
- Depth





Soil Texture



Protect Soil Structure

- Never till wet soil
- Don't "overtill' soil
- Keep heavy equipment off of soil
- Limit foot traffic in areas to be planted



Improving Soil Structure

- Coarse sand
- Vermiculite, perlite
- Manure
- Leaf mold
- Compost



Physical Factors Influencing Plant Growth



Plant Nutrients

- 17 nutrients for growth
 - ~ C, H, O from air and water
 - ~ Other 14 nutrients from soil
 - Macro-nutrients
 - Micro-nutrients



Supplying Plant Nutrients

Nutrients plants obtain from the soil

- Macronutrients:
- (needed in large amounts)
- Nitrogen (N)
- Phosphorus (P)
- Potassium (K)
- Calcium (Ca)
- Magnesium (Mg)
- Sulfur (S)

- Micronutrients:
- (needed in small amounts)
- Chlorine (CI)
- Cobalt (Co)
- Copper (Cu)
- Iron (Fe)
- Manganese (Mn)
- Molybdenum (Mo)
- Nickel (Ni)
- Zinc (Zn)

Nitrogen (N)

- Very mobile in soil
- Mobile in the plant
- Essential for leafy top growth.
- Excess N: succulent growth, dark green color, spindly plants, reduced fruiting
- N Deficiency: reduced growth, yellowing



http://www.ksuturf.com/Turf%20Diagnostic %20Guide.html

Phosphorus (P)

- Not mobile in soil
- Mobile in the plant
- Essential for root and fruit production.
- Excessive P: micronutrient deficiencies
- P deficiency: reduced growth, purpling or browning.



Phosphorus deficiency symptoms in tomato. (Epstein and Bloom 2004)

Potassium (K)

- Essential for cold hardiness, disease resistance, and stalk strength
- Excessive K: can cause Ca and Mg deficiencies
- K deficiency: reduced growth, shortened internodes, "scorched' leaf margins



Potassium deficiency symptoms in tomato. (Epstein and Bloom 2004)

Magnesium (Mg)

- Mobile in soil
- Mobile in the plant
- Excessive Mg: inhibits
 Ca uptake
- Mg deficiency: reduced growth, chlorosis, cupped leaves.





Mo

Ca

• Generally not mobile in soil or plants.

Mn

Zn

Fe

CI

Zn

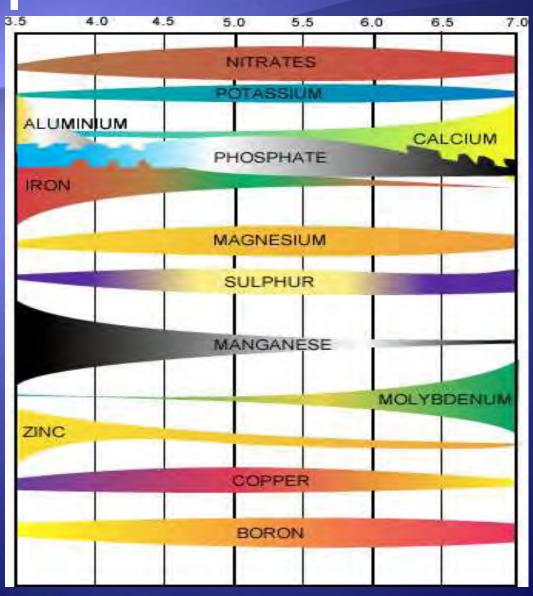
- Deficiency or toxicity problems rarely occur under normal circumstances.
- Most PA soils have adequate micronutrients for plant growth.
- Deficiencies usually related to soil pH.

Soil pH

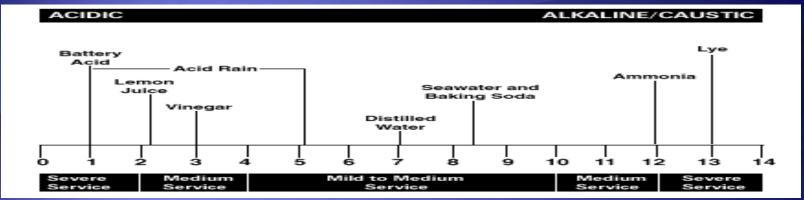
- Measures concentration of H ions
 - in soil (acidity or alkalinity)
- Ranges 0 (extremely acidic) to 14 (extremely alkaline); 7 neutral.
- Logarithmic scale
- Most plants optimum pH 6.0 to 7.0



Soil pH and Nutrient Availability



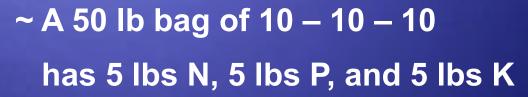
Soil pH



- Most soils in PA are naturally acidic due to:
 - Leaching of positive cations (i.e. Ca and Mg)
 - Addition of N fertilizers or manure
 - Acid rain
- To increase pH, apply lime
- To decrease pH, apply sulfur, gypsum, or other commercial products used to acidify soil.

Fertilizers

- Lists the 3 primary
 macronutrients: N P K
- Fertilizer analysis:
 - ~ % by Wt. of element in a fertilizer



~ Fertilizer Ratio: 1 – 1 – 1, 1 – 2 – 1, etc.





Fertilizers

- Complete vs. Blended fertilizer
- Special purpose fertilizers
 - Rhododendron/Azalea Food
 - Rose Food
 - Research proven?
- Slow release fertilizers
 - Dissolve slowly
 - organic fertilizers
 - Coated granules



Inorganic, Synthetic Organic & Natural Organic Fertilizers

- Inorganic made from various salts or minerals
 - Rock phosphate, potassium chloride
- Synthetic Organic man made materials
 - Urea, Ammonium Nitrate
- Natural Organic derived from living organism
 - Various by-products, blood meal, bone meal, composts, Milorganite, etc.
- Must contain a fertilizer analysis if sold as a fertilizer, regardless of type.
- Source of nutrient is not important to the plant.

Fertilizers

- Fertilizers with pesticides "weed and feed"
- Dry vs. Liquid fertilizer
- Proper application
- Timing of fertilizer applications depends on soil

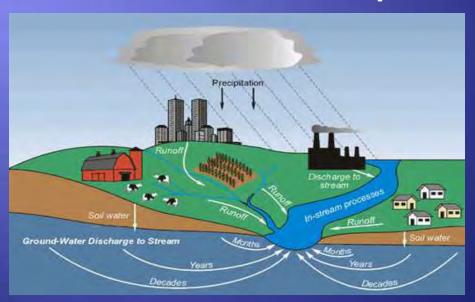
type and crop.

- Application methods
 - Broadcasting
 - Banding
 - Starter
 - Side Dressing
 - Foliar Feeding



Consequences of Over Fertilizing

- Harm to the lawn
- Pollution excess fertilizer runoff
- Affects all downstream Chesapeake Bay



Nutrients from urban, suburban and agricultural lands enter the groundwater and river flow that discharge into the Chesapeake. Once in the Bay, the overabundance of nutrients fuels the growth of algae blooms, which block sunlight and reduce dissolved oxygen levels. Image courtesy: S. Phillips / USGS

How Nutrient Management Affects the Environment

The main nutrients that contribute to pollution in our ground water and open waterways are nitrogen and phosphorus.



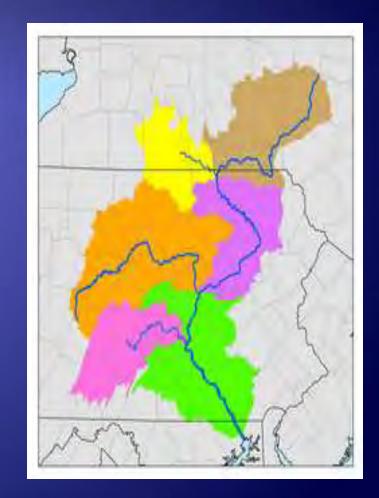
Chesapeake Bay Watershed





Susquehanna River Watershed

- As of 2000, the Susquehanna drainage basin population was 3,968,635. Its total area is 27,486 square miles
- •Accounts for 45% of Pennsylvania, 11% of New York, and 3% of Maryland.
- •The drainage basin is divided into six subbasins by the Susquehanna River Basin Commission.



Soil Testing

- Take the sample
- Fill out the form
- Review test results
- Apply indicated lime & fertilizer





(814) MG-IBEH Ppx (814) 803-45 Agricultural Analytical Services Laboratory

The Pennsylvaning State University University Park FA-16802

SOIL TE	T REPORT FO	OR:		ADDITIONA	ALCOPY TO:	
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	800-14383	12345	Centre	10		Hyblershine

SOIL NUTRIENT	LEVEL	5-1	Below Optimum	Optiman	Above Optimum
Soil pit	6.5				
Phosphorus (P)	20	ppoi	_		
Potassium (K)	80	ppm			
Magnesium (Mg)	60	ppm	_		

Limestone*: 2000 lb/A for a target pH of 6.5.

Magnesium (Mg): NONE

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Taking the Soil Sample

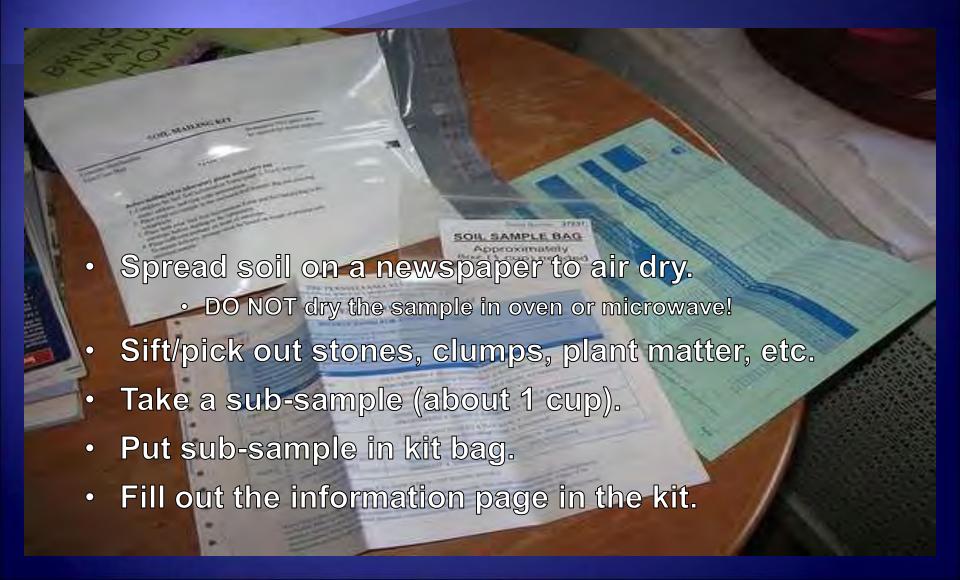
- Clean, rust-free Spade or Soil Probe
- Clean Plastic Bucket
- A Few Sheets of Newspaper
- Clean rust-free Screen or Sifter (optional)



Soil Sampling

- Don't wait until the last minute
- Sample uniform areas
- Take at least 15-20 cores from various areas of the area to be planted
- Avoid atypical areas or sample them separately
- Sample to tillage depth (or 4" for pastures)
- Collect samples in clean container, remove debris
 & let dry thoroughly
- Complete the information sheet

After Sampling....



Understanding Your Soil Test Results

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For more information on the Chesapeake Bay Program, go to http://www.chesapeakebay.net

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